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molecular structure as liquid water ( $H_2O$ ). If the bubbles were caused by the decomposition of a molecule into a gas (such as  $H_2O \rightarrow H_2$  and  $O_2$ ), then boiling would be a chemical change.

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years, scientists including Rudolf Clausius and James Clerk Maxwell developed the kinetic-molecular theory (KMT) of gases, which describes how molecule properties relate to the macroscopic behaviors of an ideal gas—a theoretical gas that always obeys the ideal gas equation. KMT

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provides assumptions  
about molecule behavior  
that can be used both as  
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where  $n$  is the molar ratio of carbon atoms to the intercalated halogens in the GIC. Upon charging,  $\text{Br}^-$  is the first species within

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the hydration layer to oxidize to a near-zero state (Br 0) and ...

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Compounds . Most covalent compounds have relatively low melting points and boiling points. While the ions in an ionic compound are strongly attracted to each other, covalent bonds create molecules that can separate from each other when a lower amount of energy is added to them. Therefore, molecular

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compounds usually have  
low melting and boiling  
points.

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