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[2] (ii) Identify the oxidizing agent in step 1..... [1] (b) A student carried out this experiment three times, with three identical small brass nails, and obtained the following results. Mass of brass = 0.456 g ± 0.001g titre 1 2 3 initial volume of 0.100 mol dm⁻³ S 2 o 3 2- (±0.05 cm)³ 0.00 0.00 0.00 Final volume of 0.100 mol dm⁻³ S 2 o 3

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- 2 - Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, HOOC?COOH·xH₂O, by titrating a known mass of the acid with a 0.100 mol dm⁻³ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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[2] (b) Marit arranged the values she found in Table 12 into an energy cycle. C₂H₄(g) + H₂(g) C₂H₆(g) 3 O 2 2CO₂(g) + 3H₂O - 1 4 1 1 k J m o l - 1 - 2 8 6 k J m o l - 1 - 1 5 6 0 k J m o l - 1 (hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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Resources for IB Chemistry SL/HL, 2017, taught by Mrs. Hala Mattar at ACS Hillingdon.

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