

## Chemistry Section 1 Review Stoichiometry Answers

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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of  $C_3H_4$ ? 2 mol O<sub>2</sub>:1 mol H<sub>2</sub>O c. What is the mole ratio ...

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Given a chemical reaction, stoichiometry tells us what quantity of each reactant we need in order to get enough of our desired product. Because of its real-life applications in chemical engineering as well as research, stoichiometry is one of the most important and fundamental topics in chemistry. Introduction to the Mole

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the space provided. 1. \_\_\_\_\_ The coefficients in a chemical equation Page 5/27

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