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Resources for IB Chemistry SL/HL, 2017, taught by Mrs. Hala Mattar at ACS Hillingdon.

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- Do not open this examination paper until instructed to do so.
- Answer all the questions.
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- The periodic table is provided for reference on page 2 of this examination paper.

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2209-6110 – 3 – Turn over M09/4/CHEMI/SPM/ENG/TZ1/XX+ 1. The molar mass of a compound is approximately 56 g mol⁻¹. Which formula is possible for this

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• The periodic table is provided for reference on page 2 of this examination paper. • The maximum mark for this examination paper is [30 marks] . 45 minutes

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When 100 cm³ of 1.0 mol dm⁻³ HCl is mixed with 100 cm³ of 1.0 mol dm⁻³ NaOH, the temperature of the resulting solution increases by 5.0 ° C. What will be the temperature change, in C, when 50 cm³

Question 3 — IB Chemistry SL — May 2017 TZ1 Paper 1 — Past IB Exams Solutions

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Exam Discussion: Chemistry SL paper 3 TZ1 : IBO

××?1 × 1000 17. Under which conditions will the reaction between 1.0 g calcium carbonate and excess hydrochloric acid be the fastest? Assume that all reactions are carried out at the same temperature. A. One large piece of calcium carbonate and 2 mol dm⁻³ hydrochloric acid B. One large piece of calcium carbonate and 1 mol dm⁻³ ...

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