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3) $2 \text{ (aq)} + 2\text{NO} \cdot 2 \text{ (g)} + 2\text{H} \cdot 2 \cdot \text{O} \text{ (l)} 3\text{Zn} \text{ (s)} + 8\text{HNO} \cdot 3 \text{ (aq)} \rightarrow 3\text{Zn}(\text{NO} \cdot 3) \cdot 2 \text{ (aq)} + 2\text{NO} \text{ (g)} + 4\text{H} \cdot 2 \cdot \text{O} \text{ (l)}$ The concentration of copper(II) ions in the resulting solution is then determined from a . calibration curve, which is plotted by measuring the light absorbance of standard solutions. Calibration curve. 0 0 00 040 00 080 100 10 000 0010 001 000 00 0030 Concentration of $\text{Cu} \cdot 2 + \text{ion} / \text{mol dm}^{-3}$

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CH₃ CH₃ CH₃ Cholesteryl benzoate (a) Identify the structural feature of cholesteryl benzoate which makes it suitable for use as a liquid crystal. [1]..... (b) Suggest the essential feature a liquid-crystal molecule must have so that the display can be turned "on" and "off". [1]..... (c) Outline the principles of a liquid-crystal display (LCD) device. [3]

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Option C - Chemistry in industry and technology 7. (a) (i) + 8 3 / 2 2 3 + ; Accept +2.7 but not +3. +2 and +3 / contains two (or more) iron ions with different oxidation states / contains Fe²⁺ and Fe³⁺; [2] Accept II and III oxidation number notation for oxidation states but not 2+ and 3+ unless ions are referred to explicitly.

CHEMISTRY STANDARD LEVEL PAPER 3

Paper 1: Multiple Choice Questions. 30 MCQs from core IB material. The use of calculators is not permitted. Written Assessment - 45 min. 20% of IB Chemistry Standard Level. Overview of assessment. The paper is assessed through a 45 min examination paper set and marked by IB. The total number of marks is 30.

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The maximum mark for this examination paper is [40 marks]. Option Questions Option A - Modern analytical chemistry 1 - 3 Option B - Human biochemistry 4 - 6 Option C - Chemistry in industry and technology 7 - 10 Option D - Medicines and drugs 11 - 13 Option E - Environmental chemistry 14 - 16

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