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strategies.***

***Chemistry Test Answers  
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Answer: Its curvy, with a  
higher bit at the end and a***

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***rather aesthetically pleasing slope downwards towards a pretty flat strait bit. The actual graph itself consists of 2 strait lines meeting at the lower left hand corner of the graph and moving away at a 90° angle. Each line has an***

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***arrow head on the end.***

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SEARCH. SUBJECTS. upper  
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Identify the letter of the  
choice that best completes  
the statement or answers the  
question. Bubble your answer***

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***on the blue side of your  
answer sheet. 1. Which step  
in the scientific method  
requires you to use your  
senses to obtain information?  
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Review ...***



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each item with the correct  
statement below. a.  
calorimeter d. enthalpy b.  
calorie e. specific heat c. joule***

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***f. heat capacity \_\_\_\_ 1.  
quantity of heat needed to  
raise the temperature of 1 g  
of water by 1°C \_\_\_\_ 2. SI unit  
of energy \_\_\_\_ 3.***

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**many of the numbers below**

**have 5 significant figures?**

**0.0054 19.000 0.00006 1.6090**

**$\times 10^8$  13607 a) 1 b) 2 c) 3 d) 4**

**e) 5 2. Nitric acid is a solution**

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***of which of the following dissolved in water? a)  $\text{HNO}_4$  b)  $\text{H}_2\text{NO}_3$  c)  $\text{HNO}_2$  d)  $\text{HNO}_3$  e)  $\text{H}_2\text{NO}_3$ . Which of the following is the smallest mass?***

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***detailed answer description,  
explanation are given and it  
would be easy to understand.***

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***positively charged.***

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an amazing subject which  
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same as the properties of the  
reactants. Different than the***



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***properties of the reactants. A combination of the properties of the reactants. Chunky substance. Valence electrons. An atom is considered to be \_\_\_\_\_ when the number of protons and electrons are equal.***

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pgs1-5. CH141 Practice Exam  
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Problems. PF answers pg1-6.  
CH141 Practice Final Key II  
(pages 6-12)***

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***In chemistry, diffusion is  
defined as the movement of  
matter by the random  
motions of molecules. In a gas***

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***or a liquid, the molecules are in perpetual motion; for instance, in a quantity of seemingly immobile air, molecules of nitrogen and oxygen are constantly bouncing off each other.***

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