

Chemistry Unit 3 Worksheet 1 Answers

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Chemistry Unit 3 Worksheet 1
Modeling Chemistry 1 U1 cp ws3 v2.0 Name Date Pd CP Chemistry – Unit 1 Worksheet 3 Mass, Volume, and Density 1. Study the matter shown in Figure 1. Each dot represents a particle of matter. [Assume the particles are uniformly distributed throughout each object, and particles of the same size have the same mass.] a.

Figure 1 B FIGURE 1 A B CP Chemistry Unit 1 Worksheet 3
As illustrated in Figure 1, the coefficients allow Equation 1 to be interpreted in any of the following ways: Two NH₄⁺ ions and one Cr₂O₇²⁻ ion yield 1 formula unit of Cr₂O₃, 1 N₂ molecule, and 4 H₂O molecules. One mole of (NH₄)₂Cr₂O₇ yields 1 mol of Cr₂O₃, 1 mol of N₂, and 4 mol of H₂O.

3.1: Chemical Equations - Chemistry LibreTexts
1 ?L = 10⁶ L . 4. millisecond – Milli (m) is the prefix and second (s) is the base unit, so the abbreviation is ms. 1 ms = 10⁻³ s . Examples. Convert 4.29 ng into grams. We are being asked to convert ng to grams so we need a metric conversion that relates the two. What does ng mean? According to the chart above, nano (n) =10⁻⁹, so 1 ng = 10⁻⁹ g

Chemistry Lesson: The Metric System & Conversions - Get ...
When converting derived units like cm³ to cubic inches, the simple conversion factor, 1 in = 2.54 cm, is applied three times in the factor-label conversion of 3 cm³ to in³.

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