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- 1. Review Notes of Mole/Gram relationship if necessary**
- 2. Complete pages 5-6 in your practice packet and check your answer.**
- 3. Complete HW#2 on google classroom and turn in by 11:59PM on Monday (3/16)**

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High School

AP Chemistry Unit – 3 : Chemical Reactions Practice Test Question 1

Which is the name of the kind of solid substance formed in this figure? A. aqueous B. precipitate C. acid D. synthesis

Question 2 An element plus additional element

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reacts to form one product is an example of which type of chemical reaction? A. combustion B. decomposition C ...

Chemistry Unit 8 Test - Chemical Reactions & Equations ...

Q. Two clear colorless solutions

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were combined. The student observed that the temperature changed from 25 °C to 23 °C when the solutions were combined and that a white substance rapidly formed and settled to the bottom of the container.

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Unit 8 : Solutions - Honors

Chemistry - Weebly

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Chemical Reactions &

Equations
Term: Definition: Found

on the right side of reaction

equations
Products
Term:

Definition: Found on the left side of

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**CHEMISTRY CP Name: KEY Period:
TEST DATE: Unit 8 Review ...**

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AP Chemistry Unit – 3 : Chemical Reactions Practice Test

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_____ Period:_____ TEST DATE:

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**Properties of Water, Solutions,
Concentration, Acids and Bases**

PROPERTIES OF WATER 1. Define

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**the following terms: polarity,
surface tension, vapor pressure,
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Covalent/molecular compound.
containing two or more nonmetals
sharing electrons. covalent bond.
bond formed when 2 electrons are
shared by two nonmetals. molecule.**

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**a molecule is the smallest particle
in a covalent compound sample.**

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Calculate the values of pH and pOH, based on K_w and the concentration of all species present in a neutral solution of water.

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Unit 8 - Moles and Stoichiometry -

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