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Key Acetate Buffer  
. 2 Acid-Base  
Titration. 5 Atomic  
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Lab. 06/23/19.  
General Chemistry  
. a student carried  
out the reaction  
below starting with  
1.993 of unknown,  
and found that the  
mass loss due to  
CO<sub>2</sub> was 0.593.

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**Mole (Unit ...**

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version of Virtual  
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installation  
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## **Bing**

Start Virtual  
ChemLab and  
select Thomson  
Cathode Ray Tube  
Experiment from  
the list of  
assignments. The  
lab will open in the  
Quantum  
laboratory. ...

Answers will vary,  
but a typical  
answer might be

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5.17 10<sup>-7</sup> m. ... A key experiment in understanding the nature of atomic structure was completed by Ernest Rutherford in 1911. He set up an experiment ...

**Answers - Model  
ChemLab Answer  
Key Acetate  
Buffer 2 Acid ...**

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This Instructor's  
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Workbook is similar  
to the Student Lab  
Manual / Workbook  
and additionally  
contains an  
overview of the full  
capabilities of the  
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version of Virtual  
ChemLab,  
installation



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instructions, and the answers for the laboratory assignments provided in the student laboratory workbook.. This product is available within:

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Activity Key Note:  
If chloride  
compounds are not  
available, metal  
nitrate compounds  
may be  
substituted. Use  
dilute or  
approximately 0.1  
M solutions  
Unknowns: Number  
the beakers 1-12.  
Add solutions as

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follows: 1.

Strontium Chloride

7. Calcium Chloride

2. Calcium Chloride

8. Strontium

Chloride 3.

Potassium Chloride

9. Barium Chloride

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Lab Stations.  
Station 1: Pop Your  
Top. 15 minutes.  
Procedure: Put on  
your goggles!  
Break the Alka-  
Seltzer tablet into  
four pieces. Fill the  
film canister most

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of the way with water. Drop in a quarter of an Alka-Seltzer tablet.

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readily identify the  
elements that had  
obvious colors  
different from all  
the others- such as  
copper that gave

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off a blue/green color and lithium that gave off a bright red color.

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In the Virtual  
ChemLab  
experiment  
"Thomson Cathode  
Ray Tube  
Experiment," which  
two of the following  
forces were  
balanced in moving  
the electron beam  
back to the center  
position: (1)  
electric, (2) strong



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nuclear, (3)

magnetic, (4)

gravity

## **chemlab questions Flashcards | Quizlet**

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designed to help students link chemical computations with authentic laboratory chemistry. The lab allows students to select from hundreds of standard reagents (aqueous) and manipulate them in a manner

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