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7. Write formulas for the following ionic compounds. Note that each includes a polyatomic ion. a) lithium phosphate b) ammonium oxide c) barium hydroxide Li_3PO_4 $(\text{NH}_4)_2\text{O}$ $\text{Ba}(\text{OH})_2$ d) calcium cyanide e) sodium chlorate f) potassium sulfate $\text{Ca}(\text{CN})_2$ NaClO_3 K_2SO_4 8. In question 3, you were asked the difference between calcium nitride and ...

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Information : Polyatomic Ions

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Advanced Ionic Bonding

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A polyatomic ion, therefore, is an ion that is made of more than one atom. An example of a polyatomic ion is the sulfate ion, SO_4^{2-} . Sulfate is composed of one sulfur atom and four oxygen atoms and overall sulfate has a negative two charge.

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chemquest - ChemQuest 19 Name Date Hour Information Naming Ions To write an ion you write the symbol of the atom and put the charge in the upper right. ... Ionic bonding is when two ions of opposite charge attract and bond to each other forming an ionic compound.

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