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16. Molecular refers to the mass of a single molecule, but molar refers to the mass of a mole of something (or the mass of  $6.02 \times 10^{23}$  atoms or molecules) . 17. 18. x atoms mol x atoms g mol g 23 23 3.82 10 6.02 10 197.0 125 x x x molecules mol x molecules g mol g 24 23 1.17 10 6.02 10 108 210 x x

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Read the Text Version. Ch. 8: ChemQuest #30 (Intro to Moles), ChemQuest#31a, #31b (% composition Divide moles of reactants by the balancing coefficient; smallest answer is LR. When aii of themembers of your group can get the answer to #7 and 8 by showing the right setup, then stapie your 6. if calcium atomsweigh 40.

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