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3.71 mol 18.53 mol. x 10/2. Percent yield = (295.3g) ÷ (333.6) ? 100 = 88.5% Consider the double replacement reaction of excess sodium chloride with 203.9g of silver nitrate. If you are able to produce 150.4g of silver chloride what was your percent yield? 203.9 g 171.98g ( theoretical yield. Divide NaCl + AgNO3 ( NaNO3 + AgCl

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