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Toluene is also called as toluol, is an aromatic hydrocarbon. It is first isolated in 1837 by distillation of pine oil by a Polish chemist named Filip Walter. It is colorless, water-insoluble and has a smell associated with paint thinners. Toluene can also be used as a recreational inhalant sometimes, and it also has the potential of causing severe neurological harm.

Toluene - Structure, Properties, Preparation and Reactions - VEDANTU

Formal charge on Cl atom of HClO_4 ion: $7 - 8/2 - 0 = 3$. Formal charge on S atom of HSO_4^- ion: $6 - 8/2 - 0 = 2$ NCERT Book Solutions. NCERT. NCERT Solutions. NCERT Solutions for Class 12 Maths. NCERT Solutions for Class 12 Physics. NCERT Solutions for Class 12 Chemistry. NCERT Solutions for Class 12 Biology.

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NCERT TEXTBOOK QUESTIONS SOLVED. 1.1. Why are solids rigid? Ans: The constituent particles in solids have fixed positions and can oscillate about their mean positions. Hence, they are rigid. 1.2. Why do solids have definite volume? Ans: Solids keep their volume because of rigidity in their structure. The interparticle forces are very strong.

NCERT Solutions For Class 12 Chemistry Chapter 1 The Solid State

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