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Thermodynamic Properties of Substances;
T3: Bond Energies The following tables
list experimental bond dissociation
enthalpies of common bonds at 298 K.

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STANDARD THERMODYNAMIC
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SUBSTANCES (continued) Molecular

$\Delta_f H^\circ / \text{kJ mol}^{-1}$ $\Delta_f G^\circ / \text{kJ mol}^{-1}$ $S^\circ / \text{J mol}^{-1}$

K^{-1} C p/J mol⁻¹ K⁻¹ formula Name Crys.

Liq. Gas Crys. Liq. Gas Crys. Liq. Gas

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Standard Thermodynamic Quantities for Chemical Substances at 25°C. Source of data: CRC Handbook of Chemistry and Physics, 84th Edition (2004).

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