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STANDARD SODIUM
THIOSULPHATE ...

Method of
Determination of
Chlorine in
Bleaching Powder

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with sodium

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Hypochlorite in
Household Bleach
I ...**

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Determination Of Available Chlorine

Solution These equations were as follows: for available chlorine, $y = 0.2723x + 0.039$ and the R^2 (determinate coefficient) = 1.000; for available iodine, $y = 1.9033x ?$

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Chem. Anal. Ed.
1935, 7, 1,
44-45

**Determination of
the
Effectiveness of
Oral Chlorine**

...

The
determination of
free chlorine in
bleach is

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possible by a redox titration.

The most common and successful method for use in high schools involves taking the sample of bleach converting the hypochlorite ion (ClO^-) to iodine (I_2) by the addition of KI

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Available Chlorine
and then
In Bleaching
Solution
titrating the
iodine with
standardized
sodium
thiosulfate
solution.

**Percent active
chlorine -
Wikipedia**

THE

DETERMINATION OF
CHLORINE IN

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In Bleaching
Sulphite

BLEACHING POWDER
BY IODINE AND
STANDARD

HYPOSULPHITE .

Apparatus ,

Reagents ,

etc. -As before.

The student may

take for

analysis a

sample of

ordinary

commercial

bleaching

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Determination Of Available Chlorine In Bleaching Solution

powder. Method
Reactions.—If,
to the milky
liquid obtained
by treating the
powder with
water, excess of
KI be added, and
then H_2SO_4 the
chlorine is
liberated from
the bleaching
powder, which

...

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estimation of

chlorine in

bleaching powder

Available

chlorine refers

to chlorine

liberated by the

action of dilute

acids on

hypochlorite.

Iodometry is

commonly

employed to

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determine the active amount of hypochlorite in bleach

responsible for the bleaching action. . . .
Determination of hydrogensulfites and sulfites

Determination of available chlorine in

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Determination Of
Available Chlorine
hypochlorite ...

method 334.0:

determination of

residual

chlorine in

drinking water

using an on-line

chlorine

analyzer office

of water (4606

m) epa

815-b-09-013

september 2009

... 6

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Available Chlorine
Determination of
Hypochlorite in
Bleach W10

**4500-CL CHLORINE
(RESIDUAL)
(2017) -
Standard Methods
For ...**

Determination of
the
Effectiveness of
Oral Chlorine
Dioxide in the

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Determination Of Available Chlorine Treatment of COVID 19 The safety and scientific

validity of this study is the responsibility of the study sponsor and investigators. Listing a study does not mean it has been evaluated by the

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Available Chlorine
U.S. Federal
Government.
Solution

ISO

**7393-3:1990(en),
Water quality ?
Determination of
free ...**

A procedure for
estimating this
interference is
available for
the DPD method.
Although the

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methods given below are useful for the determination of residual chlorine in wastewaters and treated effluents, select the method in accordance with sample composition.

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Wikipedia

Therefore, potentiometric titration is suitable for the fast and accurate determination of 1) effective chlorine/effective iodine/hydrogen peroxide content

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using the selective composite platinum (Pt) electrode or monomer platinum (Pt) indicator electrode combined with reference electrode, 2) glutaraldehyde content using the composite

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aqueous phase
In Bleaching
acid-base
Solution
electrode
(containing pH
indicator ...

**Experiment 3 -
Determination of
available
chlorine in ...**

Determination of
Sodium
Hypochlorite in
Household Bleach

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... producing the chlorine gas that we

associate with the smell of bleach. Mixing bleach with other household cleaning products is a dangerous thing to do. ... information available on the

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Determination Of
Available Chlorine
chemistry
website)
In Bleaching
Solution

**Determination of
Available
Chlorine in
Bleaching Powder**

...

The available
chlorine refers
to the chlorine
liberated by the
action of dilute
acids on the

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hypochlorite,
and is expressed
as the

percentage by
weight in the
case of
bleaching
powder.

Commercial
bleaching powder
contains 36-38
per cent of
available
chlorine. Two

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methods are in common use for the

determination of the available chlorine.

TITRATION OF ACTIVE CHLORINE WITH SODIUM THIOSULFATE

Percent active chlorine is a unit of

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concentration used for hypochlorite-based bleaches. One gram of a 100% active chlorine bleach has the quantitative bleaching capacity as one gram of free chlorine. The term "active chlorine" is

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used because
In Bleaching Solution
most commercial bleaches also contain chlorine in the form of chloride ions, which have no bleaching properties.

**ANALYSIS OF
BLEACH BY
THIOSULFATE
TITRATION**

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In Bleaching
Solution

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TITRATION OF

ACTIVE CHLORINE

WITH SODIUM

HYPOCHLORITE

SOLUTION WATER

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Determination Of Available Chlorine

AND HYGIENE 2

PREPARING THE

SOLUTIONS W The

calculation of

the quantities

required to

prepare the

differe nt

solutions

necessary for

the titration of

the active

chlorine can b e

found

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**DETERMINATION OF
CHLORINE IN
BLEACHING POWDER
BY IODINE ...**

Experiment 3
Determination of
available
chlorine in
bleach by
Iodometry, 2017
Experiment 3 -
Determination of
available

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Determination Of
Available Chlorine
chlorine in
bleach by
Iodometry Safety

Information
Bleach solution
contains
chlorine, thus
all safety
information and
risk phrases
pertaining to
chlorine must be
considered.

Safety Measures

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Determination Of
Available Chlorine
In Bleaching
Solution
Do not let it
come in contact
with the skin.

**determination of
available
chlorine in
bleaching
solution ...**

[1] B ender,
D.F., Comparison
of methods for
the
determination of

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In Bleaching
Solution

total available
residual
chlorine in
various sample
matrices, Report
No. EPA-600/4-78
-019.

Cincinnati, Ohio
45268, USA, US
Environmental
Protection
Agency (1978).

[2] Department
of the

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Determination Of
Available Chlorine
Environment.
Chemical
disinfecting
agents in waters
and effluents,
and chlorine
demand ?

**Determination of
available
chlorine in
hypochlorite ...**

The chlorine
present in the

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bleaching powder gets reduced with time. So, to find the exact quantity of bleaching powder required, the amount of available chlorine in the sample must be found out.

Chlorine will liberate free

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Determination Of
Available Chlorine
iodine from
In Bleaching
potassium iodide
Solution when
its pH is 8 or
less.

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