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Describe how iodine/thiosulfate titrations are carried out: -weigh solid oxidising agent and dissolve in water to make 250cm<sup>3</sup> (or use aqueous hydrogen peroxide as the oxidizing agent). 25cm<sup>3</sup> portions of the oxidizing agent solution are pipetted into a conical flask and some sulfuric acid is added.

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Edexcel ial chemistry unit 3 june 2018 predictions watch. And some writing parts too. - Next a reaction of halogenoalkanes or either primary or secondary alcohols with potassium dichromate and sulfuric acid. A percentage yield or atom economy que will be there I guess ( mostly % yield) All the past papers had been like this so hope this time it'll be like that too....

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Unit 3 IAL Edexcel Chemistry. Sulfur dioxide evolved and pale yellow precipitate formed after adding dilute sulphuric or hydrochloric acid.

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