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Electron Energy And Light. STUDY. PLAY. violet. high energy, high frequency, short wavelengths. red. low energy, low frequency, long wave lengths. ... for an electron to move from an energy level far from the nucleus to an energy level close to the nucleus it would need to _____ energy. lose. electron moves up.

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Electron Energy and Light 79 Extension Questions 24. The hydrogen spectral lines in Model 2 are only the wavelengths of light that are in the visible range and therefore "seen" by the naked eye. However, many other wavelengths can be detected with special equipment. a Propose a hydrogen electron transition that involves light with a ...

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Telephone Socket – Electron Energy

Energy and Electrons When an electron is hit by a photon of light, it absorbs the quanta of energy the photon was carrying and moves to a higher energy state. One way of thinking about this higher energy state is to imagine that the electron is now moving faster, (it has just been "hit" by a rapidly moving photon).

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