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compound with delocalized electrons is generally represented as a structure with localized electrons to let us know how many electrons are present in the molecule. For example, nitroethane is represented as having a nitrogen–oxygen double bond and a nitrogen–oxygen single bond.

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In this organic chemistry topic, we shall see how alcohols (R-OH) add to carbonyl groups. Carbonyl groups are characterized by a carbon-oxygen double bond. The two main functional groups that consist of this carbon-oxygen double bond are Aldehydes and Ketones.

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Explain what is meant by the term “saturated hydrocarbons.” Answer:

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12.9: Reactions of Aldehydes and Ketones with Alcohols ...

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1: Electronic Structure and Covalent Bonding. 1.1: The Structure of an Atom. 1.2: How Electrons in an Atom are Distributed.

Paula Y. Bruice | Department of Chemistry - UC Santa Barbara

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Paula Yurkanis Bruice is the author of

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two textbooks: “Organic Chemistry” (6th edition) published in 2010 by Pearson, and “Essential Organic Chemistry” (2nd edition) published in 2009 by Pearson. She is also the author of the “Study Guide and Solutions Manual” and the Companion Website that accompanies each text.

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