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## **Classifying Chem Reactions Stations**

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Reactions Page 9 ~ Classifying

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Chemical Reactions Name: Class/Lab  
Period: Classifying Chemical  
Reactions Worksheet 'Write a  
balanced chemical equation for each  
reaction and classify the reaction.

**Classifying-Chemical-Reactions-  
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## **What is a Chemical Reaction?**

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Chemistry 108 Chemical Reactions

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Prelab 4 1 Pre-Lab #4: Chemical Reactions Many chemical reactions can be placed into one of two categories: oxidation-reduction reactions and double-replacement reactions. Oxidation-reduction reactions are ones in which electrons are transferred from one species to

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Evidence of Change—ChemTopic™ Lab

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Activity, examine the chemical properties of hydrochloric acid and copper(II) chloride, identify the types of reactions that occur, and determine whether the law of conservation of mass applies to chemical reactions in a series.

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**Classifying Chemical  
Reactions—Multi-Demonstration Kit**

ChemEd13) Sharon)Geyer)

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sgeyer@pomfretschoo.org)

Classifying\*Chemical\*Reactions\*) Ada  
pted)from)Flinn%ChemTopic%Labs%  
Volume%6%

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## **Blow the Lid Off of Magnesium Oxide Lab - The Art of ...**

Student Lab Activities (259) A  
common chemical reaction used in  
chemistry class is zinc and  
hydrochloric acid. In this lab, students

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calculate how many zinc and chlorine atoms take place in the...

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## **Student Lab Activities - [flinnsci.com](http://flinnsci.com)**

Perform the demonstrations together as an introduction or as a review of reaction types, or space them over several days to focus on specific types of chemical reactions. Teacher Demonstration Notes, along with a reproducible student worksheet, are

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## **Electrochemistry—ChemTopic™ Labs Digital Collection**

Use the mass data of the magnesium before reaction and product after reaction to calculate the mole ratio of Mg and O in the compound. This mole

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ratio is the experimental chemical formula of magnesium oxide. My "go to" procedure for this lab is found in ChemTopic Labs Volume 7: Molar Relationships & Stoichiometry by Flinn. In Flinn's version of ...

## **Student Lab Activities - Flinn Sci**

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The HCl and NaOH react a calorimeter. The final temperature of the resultant solution is measured. The heat exchanged between the chemical reaction and the solution is calculated. The change in enthalpy of the reaction,  $\Delta H$ , the heat of neutralization is determined.

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