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p?V Work - Boundary Work and V?p Work

Thermal Efficiency Formula. As a result of this statement, we define the thermal efficiency, η , of any heat engine as the ratio of the work it does, W , to the heat input at the high temperature, Q_H . The thermal efficiency, η , represents the fraction of heat, Q_H , that is converted to work. It is a dimensionless performance measure of a heat engine that uses thermal energy, such as a ...

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Since the volume remains constant, the heat transfer into or out of the system does not the p?V work but only changes the system's internal energy (the temperature). The case $n = 0$ corresponds to an isochoric (constant-volume) process for an ideal gas and a polytropic process.. See also: What is an Ideal Gas. Let assume an isochoric heat addition in an ideal gas.

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p?V Work p?V Work is equal to the area under the process curve plotted on the pressure-volume diagram. Pressure-volume work (or p?V Work) occurs when the volume V of a system changes. The p?V Work is equal to the area under the process curve plotted on the pressure-volume diagram. It is also known as boundary work. Boundary work occurs because the mass of the substance within the system ...

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