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**Acid - Wikipedia**

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Nella teoria di Arrhenius gli acidi sono definiti come sostanze che, sciolte in acqua, liberano ioni  $H^+$  (ioni idrogeno). Le basi, invece, sono tutte quelle sostanze che, in soluzione acquosa, liberano ioni  $OH^-$  (ioni idrossido).. Nel 1923 il chimico danese Johannes Nicolaus Brønsted e

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l'inglese Thomas Martin Lowry  
elaborarono indipendentemente l'uno  
dall'altro una teoria più generale, che  
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## **Teoria acido-base di Brønsted-Lowry - Wikipedia**

A teoria de Brønsted-Lowry é uma teoria sobre a reação entre ácidos e bases proposta independentemente

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por Johannes Nicolaus Brønsted e Thomas Martin Lowry em 1923. O conceito fundamental proposto é que, quando um ácido e uma base reagem entre si, o ácido forma sua base conjugada, enquanto a base forma seu ácido conjugado; com a reação sendo mediada pela troca de um

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Gallium(III) trioxide is an inorganic compound with the formula  $\text{Ga}_2\text{O}_3$ . It exists as several polymorphs, all of which are white, water-insoluble

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solids. Although no commercial applications exist,  $\text{Ga}_2\text{O}_3$  is an intermediate in the purification of gallium, which is consumed almost exclusively as gallium arsenide. The thermal conductivity of  $\beta\text{-Ga}_2\text{O}_3$  is at least one order of magnitude lower ...

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above to get started. Many online periodic tables only provide basic information for a particular element.

### **Gallium(III) oxide - Wikipedia**

An acid is a molecule or ion capable of donating a proton, (i.e. hydrogen ion,  $H^+$ ), (a Brønsted–Lowry acid), or,

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alternatively, capable of forming a covalent bond with an electron pair (a Lewis acid).. The first category of acids are the proton donors, or Brønsted–Lowry acids. In the special case of aqueous solutions, proton donors form the hydronium ion  $\text{H}_3\text{O}^+$  and are known as ...

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