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12NaCl(l) 2Na(l) Cl (g)/ NaCl(l) Na(l) Cl (g) 222 ; [2] Award [1] for oxidation and reduction half-equations. Award [1] for overall cell reaction, including correct state symbols. Accept Na+(l) + Cl-(l) instead of NaCl(l) as a reactant. Penalize equilibrium arrows once only. (b) Ions not free to move when solid / Ions in rigid lattice / OWTTE; [1]

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