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Assuming salicylic acid $C_7H_6O_3$ as the limiting reagent calculate the theoretical yield. Molar mass of $C_7H_6O_3 = 132.1$ g/mol. Molar mass of

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aspirin = 180.1 g/mol

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10.3 Percent Composition and Chemical Formulas 10

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Match. Gravity. Created by. madigil33. Terms in this set (35) Figure

33 If P1 is 50 kPa and P2 is 100 kPa, what are V1 and V2? 2.0L 1.0L.

How would the graph in Figure 3-3 change if the volume of the cylinder remained constant and temperature was the manipulated variable?

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Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions:

Formation of Solids, page 345 Analysis 1. a. $\text{Na}_2\text{CO}_3 + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + 3\text{Ag}_2\text{CO}_3(\text{s})$ b. $2\text{Na}_3\text{PO}_4 + 3\text{Pb}(\text{NO}_3)_2 \rightarrow 6\text{NaNO}_3 + \text{Pb}_3(\text{PO}_4)_2(\text{s})$

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Chapter 11 Chemical Reactions Interpreting Graphics 1 . Name Date

Class carbonic acid _____ nitro benzene nitric acid ... 2 Chapter 11

Chemical Reactions Interpreting Graphics. Name Date Class CHEMICAL

REACTIONS Vocabulary Review Each clue describes a vocabulary term.

Read the clues and write the letters of each term on the lines

provided.

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1.1 Chemistry 1

Chapter 13 Interpreting Graphics and Vocabulary Name . Part A – Intermolecular Forces. 1. Fill in the diagram (with high or low) to show how intermolecular forces influence the volatility, vapor pressure, and boiling point of a substance.

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25.2 Nuclear Transformations 25

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Introduction to Chemistry 7 1.1 FOCUS Objectives 1.1.1 Identify ?ve traditional areas of ... Review, Interpreting Graphics ... 8 Chapter 1 Areas of Study Because the scope of chemistry is vast, chemists tend to focus on one area.

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School. Name Date Class INTERPRETING GRAPHICS Use with Section 3.2
Figure 1 Use Figure 1 to answer the following questions. 1.

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Review, Interpreting Graphics ... chem_TE_ch25.fm Page 803 Thursday,
April 20, 2006 8:47 AM. 804 Chapter 25 Section 25.2 (continued) Half-
Life Discuss Explain that, for each element, there exists only a small
range of neutron-to-proton ratios that produce stable nuclei. If a
nucleus does not reflect a

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