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1.1.2.1 Derived units from the SI system 1.1.3  
CGS (cm-g-s) system***

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***Independent and dependent chemical reactions; Inerts versus Reactive Species; Equilibrium constants (introduction/review from general chem) Extent of Reaction is still Extent of Reaction; Example Problem without equilibrium; Example Problem with equilibrium; Reactions with Recycle. Example of how a separation process can be used to improve efficiency.***

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***approach a problem? Problem Type: What is the underlying mechanism/theory? Use: Scientific Method Define the Problem Research what's been done before Lots of thinking Form a hypothesis (testable prediction) Test Hypothesis (actual experiment)***

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***that of the first, and the ratio of moles water vapor to moles of water is the same in both tanks.***

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### ***Introduction To Chemical Engineering Problems***

#### ***What are Chemical Engineers?***

***“Chemical engineers” use math, physical sciences (physics, chemistry), life sciences (biology, microbiology and biochemistry), and economics to solve practical problems. The difference between chemical engineers and other types of engineers is that they apply a knowledge of chemistry in addition***

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(Cambridge Series in ...***

***LECTURE 10. Prof. Manolito E Bambase Jr.***

***Department of Chemical Engineering. Example 11-1. In a given process, 100 kmol of carbon is burned in a furnace. It has been found that 20% of the carbon undergoes incomplete combustion resulting to CO production. The rest of the carbon undergoes complete combustion.***

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***Equation for each component which is called (component material balance). If there is no chemical reaction the number of equations that can be written is equal to the number of components in the system. c. Equation for each element which is called (element material balance) if there is a chemical reaction.***

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