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electromagnetic waves. amplitude amplitude A B Interpreting Waves MATH
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substance C H C H C H O O O CO CO CO H O H O H O number of moles of
unknown substance O CO H O C H CO H O C H O H O C H O CO use mole
ratio 5 mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1
mol ...*

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Determine the molar mass of each of the following compounds.*

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• Answer questions on Chapter 3 in preparation for the test. Math Skills Transparency 4 and Master TWE, p. 76 TWE, pp. 54–85 15 minutes Suggested Activity Resources Time Lesson Plan Homework • Have students complete Section 3.4 Assessment. • Assign relevant questions from Chapter 3 Assessment.

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42 82 1. In equation 1, what do the numbers 238 and 92, written to the left of the symbol U, represent? 238 is the mass number of the isotope, and 92 is the atomic number of the isotope. 2. Explain how both mass number and atomic number have been conserved in equation 1.

Matter—Properties and Changes - Glencoe

Calculating Numbers of Electrons Use with Chapter 7, Section 7.1 and Predicting Ionic Change 1. What happens to a neutral atom if it loses one electron? Why? AS 2. What happens to a neutral atom if it gains two electrons? Why? 3. Write a simple word equation that shows how you can figure out the charge of an ion, ... Math Skills Transparency ...

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