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Answer Key

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Molecular Theory The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties.

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CHAPTER 5
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SHORT ANSWER 3

Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to

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increasing atomic
number. (d) based on
when they were
discovered. 2. d
Mendeleev noticed
that certain
similarities in the ...

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Answer the following
questions in the space
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and we are in Chapter
13. I need the answer
to review concept
number 10. Where can
I find the answer
online? Ok so
basically, I am
stumped on Number
10 of the review
concepts at the end of
the chapter, chapter
13. The name of my

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there a web site that

has the answers that I
can access?

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States Of Matter

CHAPTER 10

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Chapter 10 Review

Answer: Short Answer

Answer the following

questions in the space

provided. 1. Indicate

whether each of the

following is a physical

or chemical property

of water. _____ a. The

density of ice is less

than the density of

liquid water. _____ b.

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Integrated Science
End-of-Chapter
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Answers to Chapter 1
Review Questions 1
The era of modern
science in the 16th
century was launched
when Galileo Galilei
revived the

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concept number 10.

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Chapter Test Name

Class Date Chapter

Test B, continued 27.

Distinguish between
ionic crystals and
metallic crystals.

PART V Write the
answers to the
following questions on
the line to the left, and
show your work in the

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space provided. The
molar enthalpy of
fusion for water is
6.008 kJ/mol. 28.

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Change Concepts . CH
10 Section Review 13.1
& 13.2 . CH 10 Section
Review 13.3 & 13.4 .
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Characteristics of

Gases MIXED

REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1.

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SHORT ANSWER

Answer the following
questions in the space
provided. 1. Match

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description on the right to the correct crystal type on the left.

b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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