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Building Blocks of Matter ... exactly 12 g of carbon-12 is referred to as a(n) ...

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continued b. Explain the difference between an open system and a closed system. 5.

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Chapter 6: Chemical Bonding, Modern Chemistry. This is a working presentation of the notes for this chapter. Meaning that we may or may not cover all of the material here.

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numbers 1246120, 1525057, and 1413739.

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Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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1 amu, or is exactly  $1/12$  the mass of a  
carbon-12 atom. average atomic mass. is  
the weighted average of the atomic masses  
of the naturally occurring isotope of an  
element. ... Modern Chemistry Chapter 3  
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space provided, write the letter of the term

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or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas
- a. attract each other but do not collide.
  - b. repel each other and collide.

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Modern Chemistry Chapter 12-Solutions

Section 1- Types of Mixtures Solutions are  
homogeneous mixtures of two or more  
substances in a single phase. Soluble

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describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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Chapter Test Chapter Test B, continued

16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.

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