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CHAPTER 6 REVIEW Chemical Bonding MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. a.

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Modern Chemistry Chapter 6 Chemical Bonding . Chemical Bond -A link between atoms that results from the mutual attraction of their nuclei for electrons -Electrostatic attraction between proton and electron -Classified by the way the valence e- are distributed around nuclei of combined atoms .

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. _____. In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms.(d) involved in covalent bonds. 2.

CHAPTER 6 REVIEW Chemical Bonding

The Chemical Bonding chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with chemical bonding.

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Intermolecular forces do NOT have effect on _____. solubilityacidityevaporationboiling pointQuestion 33. What is the correct definition of an ionic bond. a type of ...

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Modern Chemistry Chapter 6. Lessons: -Introduction to Chemical Bonding -Covalent Bonding and Molecular Compounds -Ionic Bonding and Ionic Compounds -Metallic Bonding -Molecular Geometry. STUDY. PLAY. chemical bond. a mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together.

Modern Chemistry Chapter 6

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Chapter 6 – ModernCHEMICAL BONDING Chemistry S.Martinez 2. CHEMICAL BONDING In a chemical bond between atoms, their valence electrons are redistributed in ways that make the atoms more stable. The way that atoms are redistributed determines the type of bonding.11/19/2010 S.Martinez 2

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