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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass numberand the atomic numberof a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope. Atomic number is the total number of protons ...

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Modern Chemistry Chapter 8. STUDY. PLAY. chemical equation (8.1) a way to describe a chemical reaction using chemical formulas and other symbols. precipitate (8.1) a solid that forms from a solution during a chemical reaction. coefficient (8.1) A number in front of a chemical formula in an equation that indicates how many molecules or atoms of each reactant and product are involved in a ...

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18. The valence electron configuration for the Group 16 ...

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

3 Atoms: The Building Blocks of Matter - Ms. Sheehan's ...
Modern Chemistry • CHAPTER 8 HOMEWORK 8-3 (pp. 248 – 250) VOCABULARY Complete each sentence. 1. A chemical equation is like an algebraic equation because they both _____. 2. The coefficients of a chemical reaction indicate _____. 3.

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How It Works: Identify the lessons in Holt McDougal Modern Chemistry's States of Matter chapter with which you need help. Find the corresponding video lessons within this companion course chapter.

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Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A, continued _____ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of their atomic number. c. wave patterns repeat at regular intervals.

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Accelerated Chemistry Anderson - MCHS Modern Chemistry 3 Chemical Equations & Reactions CHAPTER 8 STUDY GUIDE Chemical Equations and Reactions SECTION 8-1 SHORT ANSWER Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right.

HOMEWORK 8-1
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