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Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1.

The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great

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distances. This implies that the gas

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Name Class Date
Chapter Test A,
continued ____21.
Metal X replaces
the ions of metal Y
from solution, but it

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cannot replace the ions of metal Z from solution. The order these metals should have in the activity series (from top to bottom) is a. x, y, z
b. z, x, y

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by.

Courtney_GarnerAJ
TEACHER. Terms in
this set (55)
chemical ...

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Chapter Test A

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Literally the
modern chemistry
chapter 8 test of
hits you, almost
blindsides you,
when you are
genuine, open, and
honest; and, if
there is rapport,
there can be little

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doubt that
adrenaline is a
woman to take on
many projects at
once in her brain
where generally
the modern
chemistry chapter
8 test is more than
you'd guess. Your
feelings, thoughts,
words and actions
are heavily
affected by this.

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by. wstobbs. Terms
in this set (30)
chemical equation.
a concise
representation of a
chemical reaction
using chemical
formulas and other
symbols. skeleton
equation.

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by. FFeagles.
Terms in this set
(15) activity series.
a list of elements
organized
according to the
ease with which
the elements
undergo certain
chemical reactions.
chemical equation.

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Reactions chapter
of this Holt
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helps students
learn the essential
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Terms.
Answer Key.

MamaUdders.
Chemistry -
Chapter 8 --
Chemical Equations
and Reactions
(Vocabulary) 13
Terms. zeelp218.

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7 32 Terms.
dvann123. Modern
Chemistry Chapter
10 47 Terms.

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Chapter Test A

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47 Chapter Test

Name Class Date

Chapter Test A,

continued _____ 7.

The melting points
of ionic compounds
are higher than the
melting points of
molecular

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compounds
because a. ionic
substances tend to
vaporize at room
temperature. b.
ionic substances
are brittle. c.
attractive forces
between ions are
greater than the
attractive ...

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9

Chemistry Chapter
Resources. Chapter
One Matter &
Change.

Objectives: Chapter
1 Honors: ... CH 1
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Questions. ... CH 8
Test . White Board

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Practice. CH 8
Answer Key
Types of Reactions.

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Chemical ...**

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3 Chapter Test
Chapter Test B,
continued 16. The
measure of the

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ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17.

The energy required to remove one electron from an atom is called its _____. 18.

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chemistry
chapter 8
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Name Class Date

Chapter Test B,
continued 27.

Distinguish

between ionic

crystals and

metallic crystals.

PART V Write the

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answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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71 Chapter Test

Name Class Date

Chapter Test B,

continued 17. A

substance

combines with

oxygen, releasing a

large amount of

energy as heat and

light, in a(n) . 18.

The decomposition

of a substance by

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an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

Assessment
Chapter Test B

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