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CHAPTER 6 REVIEW Chemical Bonding SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

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Chapter 9 – Chemical Calculations and Chemical Formulas 119 Chapter 9 Map Chapter Checklist Read the Review Skills section. If there is any skill mentioned that you have not yet mastered, review the material on that topic before reading this chapter. Read the chapter questions before the lecture that describes it.

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5 The Periodic Law

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Assessment Chapter Test B

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

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Modern Chemistry 92 Chapter Test Name _____ Class _____ Date _____ Chapter Test B, continued 31. Calculate the energy released by freezing 13.8 g of liquid. Its molar mass is 82.9 g/mol, and its molar enthalpy of fusion is 4.60 kJ/mol. PART VI Define the term equilibrium vapor pressure and use it to explain the following two phenomena.

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6 Chemical Bonding

Chapter 6: Chemical Bonding, Modern Chemistry. This is a working presentation of the notes for this chapter. Meaning that we may or may not cover all of the material here.

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