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Chapter 1 Practice Test B | Chemical
Elements | Chemical ...

Modern Chemistry 11 . Name Section
Quiz, continued Class Date 6. A

measurement is accurate if it a. is
reproducible. b. close to the true value. c.
has many decimal places. d. has many

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significant figures. 7. To determine the number of significant digits in a measurement, follow

Assessment Chapter Test B

Modern Chemistry 71 Chapter Test Name

Class Date Chapter Test B, continued 17.

A substance combines with oxygen,

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releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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CHAPTER 3 TEST continued Date Class

FILL IN THE BLANK Write the correct

term (or terms) in the space provided. 9, If

a particular ompound is composed of

elements A and B, the ratio of the mass of

B to the mass of A will alw ys be the same

This is a statement of the law of exactly 1

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g of carbon-12 is referred to as a(n) 11.

3 Atoms: The Building Blocks of Matter -
Ms. Sheehan's ...

Chapter 1 Practice Test B Chapter:

Matter and Change _____ 1. Chemistry is
a natural science that deals with the study
of a. living things and their life processes.

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b. the physical features of Earth. c. the composition, structure, properties, and changes of matter. d. the composition, motion, and relative positions of stars and planets. _____ 2.

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Holt Chemistry 40 Matter and Energy
Section: Measurements and Calculations
in Chemistry In the space provided, write
the letter of the term or phrase that best
answers the question. _____ 1. A
measurement is said to have good
precision if it a. agrees closely with an
accepted standard. b. agrees closely with

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other measurements of the same quantity.

Chemistry Chapter Test B Answer Key
Flashcards | Quizlet

Modern Chemistry 111 Chapter Test
Name Class Date Chapter Test B,
continued 18. Dissolution processes with
negative enthalpies of solution are

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processes. 19. is the solution process with water as the solvent. 20. Liquid solutes and solvents that are not soluble in each other are. 21.

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Modern Chemistry 141 Chapter Test

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Name _____ Class _____ Date _____ Chapter Test B,
continued 18. When a strong acid is
titrated with a weak base, the pH of the
solution at the equivalence point is than 7.
19. A is a highly purified solid used to
check the concentration of a standard
solution. 20. A 1 M solution of NaOH will
have a pH that is

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Assessment Chapter Test B - Baumapedia
Modern Chemistry 71 Chapter Test Name
Class Date Chapter Test B, continued 17.
A substance combines with oxygen,
releasing a large amount of energy as heat
and light, in a(n). 18. The decomposition
of a substance by an electric current is

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Modern Chemistry 132 Chapter Test

Name Class Date Chapter Test B,

continued Refer to the following statement

to answer questions 30–32: Dilute

HCl(aq) and NaOH(aq) are mixed in

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chemically equivalent quantities. 30. Write the chemical equation for the reaction. 31. Write the overall ionic equation for the reaction. 32. Write the net ionic equation.

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b. Explain how an atom can exist in this

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state. Atoms consist of a positively charged nucleus, made up of protons and neutrons, that is surrounded by a negatively charged electron cloud. The positive and negative charges combine to form a net neutral charge. MODERN CHEMISTRY ATOMS: THE BUILDING BLOCKS OF MATTER 19

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Modern Chemistry Chapter Test B
Modern Chemistry 88 Chapter Test
Chapter: States of Matter PART I In the
space provided, write the letter of the term
or phrase that best completes each
statement or best answers each question.

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1. According to the kinetic-molecular theory, particles of an ideal gas
a. attract each other but do not collide. b.
repel each other and collide.

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Holt McDougal Modern Chemistry 3

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Chapter Test Chapter Test B, continued
16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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PLAY. Match. Gravity. Created by.
jkinsler12. Gases and Gas Laws. Terms in
this set (26) Ideal Gas Law. $PV=nRT$. P.
Pressure. V. Volume. n. ... temperature
and pressure are inversely proportional to
the square roots of their molar masses
Rate of effusion of A / B = $\sqrt{M_b / M_a}$...

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Assessment Chapter Test B - Ed W. Clark
High School

Modern Chemistry 86 Chapter Test Name
Class Date Chapter Test A, continued

_____13. The separation process of paper chromatography can be explained by a.
vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of

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gravity. _____ 14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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