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Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

Modern Chemistry Chapter Atoms Test Answers

Modern Chemistry Chapter Test Modern Chemistry 101  
Chapter Test Name Class Date Chapter Test B, continued  
18. The pressure exerted by each gas in a mixture is called the of that gas. 19. If the temperature and number of moles of

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a gas remain constant but the volume increases, the pressure of the gas will . 20.

Modern Chemistry Assessment Chapter Test Reaction Kinetics

Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.

Assessment Chapter Test B

MODERN CHEMISTRY CHAPTER 1 TEST Matter and

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Change MULTIPLE CHOICE On the line at the left of each statement, write the letter of the choice that best completes the statement or answers the question. 1) Chemistry is a natural science that deals with the study of.

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MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11  
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ANSWER KEY 1 Modern chemistry answer key chapter 2. a

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2. b 3. c 4. c 5. a 6. b 7. Modern chemistry answer key chapter 2. c 8. Polarity in a water molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic.

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