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In modern theory, it is now referred as coordination number of central metal atom or ion. 9.2 FeSO_4 solution mixed with $(\text{NH}_4)_2 \text{SO}_4$ solution in 1:1 molar ratio gives the test of Fe^{2+} ion but CuSO_4 solution mixed with aqueous ammonia in 1:4 molar ratio does not give the test of Cu^{2+} ion. Explain why. Ans. FeSO_4 solution mixed with $(\text{NH}_4)_2 \text{SO}_4$ solution in 1 : 1 molar ratio forms a double ...

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NCERT TEXTBOOK QUESTIONS SOLVED. 1.1. Why are solids rigid? Ans: The constituent particles in solids have fixed positions and can oscillate about their mean positions. Hence, they are rigid. 1.2. Why do solids have a definite volume? Ans: The constituent particles of a solid have fixed positions and are not free to move about, i.e., they possess rigidity.

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