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Boric acid - Wikipedia
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The HITRAN2016 molecular spectroscopic database
First Annual Law School Fair: coronavirus style. @uark.prelawsociety it's been great being your president, but I swear I've seen it all at this point! (disclaimer for anyone seeing this post-2020, this event happened the day before my university shut down.

Stocks - Bloomberg
How many tons of each of the later two samples should be mixed with the original 10 tons to give 20 tons sample containing 1.7% sulphur? 9. CS 2 and Cl 2 in the weight ratio 1 : 2 are allowed to react according to equation: CS 2 + 3Cl2 > CC14 + S2C12, calculate: (a) Which reactant is limiting reagent.

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Examples of nucleophiles are anions such as Cl-, or a compound with a lone pair of electrons such as NH3 (ammonia). In the example below, the oxygen of the hydroxide ion donates an electron pair to form a new chemical bond with the carbon at the end of the bromopropane molecule.

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7. A sample of soil is mixed with water and allowed to settle. The clear supernatant solution turns the pH paper yellowish-orange. Which of the following would change the colour of this pH paper to greenish-blue? (a) Lemon juice (b) Vinegar (c) Common salt (d) An antacid. 8. Which of the following gives the correct increasing order of acidic ...

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Hydrogen chloride - Wikipedia
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JEE Main Chemistry Notes, JEE Main Chemistry Syllabus
Boric acid, also called hydrogen borate, boracic acid, and orthoboric acid is a weak, monobasic Lewis acid of boron.However, some of its behaviour towards some chemical reactions suggest it to be tribasic acid in the Brønsted sense as well. Boric acid is often used as an antiseptic, insecticide, flame retardant, neutron absorber, or precursor to other chemical compounds.

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Hydrogen chloride is a diatomic molecule, consisting of a hydrogen atom H and a chlorine atom Cl connected by a polar covalent bond.The chlorine atom is much more electronegative than the hydrogen atom, which makes this bond polar. Consequently, the molecule has a large dipole moment with a negative partial charge (δ-) at the chlorine atom and a positive partial charge (δ+) at the hydrogen ...

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