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Key Takeaways Key Points. A balanced nuclear equation is one where the sum of the mass numbers (the top number in notation) and the sum of the atomic numbers balance on either

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side of an equation. Nuclear equation problems will often be given such that one particle is missing.

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Where To Download Nuclear Equation Practice Answer Key decay of  $^{223}\text{Fr}_{87}$ .  $^{223}\text{Fr}_{87} \rightarrow ^{223}\text{Ra}_{88} + ^4\text{He}_2 + e^-$  3. Write a nuclear equation for the alpha and beta decay of  $^{149}\text{Sm}_{62}$ .  $^{149}\text{Sm}_{62} \rightarrow ^{145}\text{Pm}_{61} + ^4\text{He}_2 + e^-$  4. Nuclear Equation Practice Tesccc Key Chemistry Nuclear equations worksheet answers typepad pages 1 3

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alpha and beta decay

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chemistry as ... Write the nuclear equation for this reaction and identify the two other particles.  ${}_{92}^{235}\text{U} \rightarrow {}_{40}^{137}\text{Zr} + {}_{52}^{92}\text{Te} + 2\text{n}$  ?  
Te+ Zr+2 The two other particles formed are neutrons.

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Nuclear Equations

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