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Anything that has definite mass and volume; anything that take... A covalent bond in which two pairs of electrons are shared. A chemical bond formed by the sharing of electrons. Group of non-metal atoms held by covalent bonds. A molecule that consists of two atoms of the same element.

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a chemical equation that does not indicate the relative amounts of the reactants and products. catalyst a substance that speeds up a reaction but is not used up in the reaction and is neither a reactant or a product.

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Guided Reading And Study Workbook Chapter 10
11.2 Types of Chemical Reactions> 30 A halogen can also replace another halogen from a compound. •The activity of halogens decreases as you go down Group 7A of the periodic table—fluorine, chlorine, bromine, and iodine. •Bromine is more active than iodine, so this reaction occurs: Br 2 (aq) + 2NaI(aq) 2NaBr(aq) + I 2 (aq)

Chapter 12 Stoichiometry Answer Key Pearson
A chemical reaction can be concisely represented by a chemical 1.. The substances that undergo a chemical change are the 2.. The new substances formed in a chemical reaction are the 3.. In accordance with the law of conservation of . 4. a chemical equation must be balanced. When balancing an 5. equation, you place in front of reactants and products so 6.

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