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Chapter 1

Introduction: Matter
and Measurement

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Quinnipiac University

Hamden, CT Lecture

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Quizlet

The melting of ice to
water is an example

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of a physical change, and the burn- ing of wood is an example of a chemical change.

A mixture is made of two or more substances. A

substance is made of only one type of matter. An element is a pure substance that cannot be bro- ken down by chemical means.

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Chapter 3 Students
will understand how
the elements are ...

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Flashcards. Learn.

Write. Spell. Test.

PLAY. Match. Gravity.

Created by.

Rachel_Frishberg.

Terms in this set (8)

Nucleus. The very
small center core of

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an atom. It contains
the protons and
neutrons. Protons.

Located inside of the
nucleus. Have a
positive (+) electric
charge.

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Smallest unit of an

element that

maintains the

properties of that

element. John Dalton.

Chemist & teacher

who wanted to know

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why elements combine in certain sizes based on the mass to form compound. His result said that elements combine in specific proportions because theyre made of single atoms.

McMurry & Fay,
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A molecule is a particle formed when two or more atoms are held together by chemical bonds. A compound is a pure substance made of two or more elements chemically combined in a set ratio. A chemical formula shows the elements in the compound and the ratio of atoms.

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Atoms, Molecules,
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and Reactions: An
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atoms of the same
element that differ in
the number of
neutrons, but have
the same number of
protons d. the particle
of an atom that is
negatively charged e.
a specific amount of
energy related to the
movement of

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f. electrons in atoms
g. the particle of an atom with a positive charge
h. the number of protons in the nucleus of every atom of an element

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This section describes
the structure of atoms

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and how atoms are
related to elements.

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on the chemical

processes that

underlie critical

biological functions.

1.1 The Scope of

Chemistry >

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Understanding Ideas

1. Name three

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particles found in an atom. 2. Which two particles are found in an atom's nucleus? 3. An atom has the same number of which two particles? 4. How many protons are in a carbon atom? ...

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An atoms-first approach to chemistry: Through an atoms-first approach, used effectively in the previous four editions, students start learning from the building blocks of

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matter and progress
to understanding
complex chemistry
concepts from a
logical point of view
and with a deep
understanding.

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chapter 3, section 1:

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Chemistry: Atoms

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Chemistry Slides.ppt

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