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Terms in this

set (32)

Elements are

organized in the

\_\_\_\_\_ by

increasing

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atomic number.  
periodic table.  
In the late  
1800's, Dmitri  
Medeleev devised  
the first  
periodic table  
based on \_\_\_\_\_.

## **6.3 Periodic Trends - Henry County School District**

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Section 6.1

Development of

the Modern

Periodic Table

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1. octaves
2. eight
3. nine
4. accepted
5. Dmitri Mendeleev
6. atomic mass
7. elements
8. atomic number
9. Henry Moseley
10. protons
11. periodic law

**Directed Reading  
For Content  
Mastery Section**

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## 3 The ...

The Periodic  
Table and  
Periodic Law  
Section 6.1  
Development of  
the Modern  
Periodic Table  
... Use the  
periodic table  
on pages 156–157  
in your textbook  
and the periodic  
table below to

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answer the

following

questions. s

block He p block

d block dl "29'

... Section 6.3

Periodic Trends

**Periodic table  
section composed  
of group 3-12 -  
Answers**

Organizing the  
Elements

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Periodic Table-  
elements are  
arranged by  
increasing  
atomic number  
and by changes  
in physical and  
chemical  
properties.

Mendeleev had to  
leave blank  
spaces in his  
periodic table  
to keep the

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Section 3 Answer  
Key

elements

properly lined  
up according to  
their chemical  
properties.

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**Reinforcement**  
**The Periodic**  
**Table Worksheet**

...

Section 3.2  
Answers and  
questions of the

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answers from our  
randork homework  
to study by.

Terms in this  
set (21)

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Elements. the  
periodic table  
group  
similar\_\_\_\_\_  
together?  
properties. When  
grouped  
together, it is  
easier to  
predict ...

**Chapter 3 Atoms,  
Elements, and  
the Periodic**



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Chapter 3 -  
Elements and the  
Periodic Table  
Study Guide ...  
How can you  
easily tell the  
answer to #3 by  
looking at the  
periodic table?  
Their ... III.  
Patterns of the  
Periodic Table  
(45 points on

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test) This section is the largest of the test - so focus on it!! 1. What are the rows called?

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Section 5.1: The  
History of the  
Periodic Table

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Key  
Dmitri Mendeleev  
(1869) - first  
person to  
organize the  
elements in a  
chart Organized  
about 70  
elements by  
increasing  
atomic mass Left  
blank spaces for  
elements which  
were not  
discovered yet

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key section  
answer. ...

Section 3  
Reinforcement  
Periodic Table

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**ANSWERS**

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Periodic Table  
Notes • John  
Newlands  
proposed an  
organization  
system based on  
increasing ... o  
Lanthanides (4f  
section) Also  
called the rare

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earth metals,  
because they're  
rare. Usually  
intermediate in  
reactivity  
between alkaline  
earth metals and  
transition  
metals.

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Table; Section 3  
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form of and can

**Unit 3 Notes:**  
**Periodic Table**  
**Notes**

The Periodic Law  
Section Quiz:

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Electron

Configuration

and the Periodic

Table In the

space provided,

write the letter

of the term or

phrase that best

completes each

statement or

best answers

each question.

25. 1. An

element that has

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Section 3 Answer  
Key

the electron  
configuration  
[Ne]3s 3p 1s in  
Period 25. 2.

**Chapter 5: The  
Periodic Law**

Section 6.3 6.3  
FOCUS Objectives  
6.3.1 Describe  
trends among the  
elements for  
atomic size.  
6.3.2 Explain

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how ions form.

6.3.3 Describe periodic trends for first ionization energy, ionic size, and electronegativity.

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compare/contrast  
table for cation  
and anion, and  
use the ...

## **Section 3.2 Answers and questions of the periodic table**

...

Atomic structure  
chemical bonds  
bond 6 4k views  
reading

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The Periodic Law  
SECTION 3 SHORT  
ANSWER Answer  
the following  
questions in the  
space provided.

1. c When an  
electron is

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Key

added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an



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electron from a  
neutral atom is  
the atom's

**Section 3 The  
Periodic Table -  
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Notes - Section  
2. Elements -  
matter made up  
of only one kind  
of atom. There  
are 115 known

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elements. 90 naturally occurring elements, plus synthetic elements - made by scientists. Periodic Table - Chart that organizes and displays information about the elements. Atomic

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Number - the top  
number in the  
element's  
periodic table  
block

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...

nonmetal more  
family group  
left periodic

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table metal  
metalliod period  
right properties  
atomic number 1.  
The chart that  
lists the  
elements in an  
organized way is  
called the  
\_\_\_\_\_. 2. The  
periodic table  
lists the  
elements  
according to

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**Section 3**  
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**Answers**

Gold (Au) is a transition metal, thus is positioned in the d-block (middle portion) of the main

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section of  
periodic table  
in Period 6,  
Group 11 (or  
Group I b).

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Properties of  
atoms and the  
periodic table  
atomic structure  
chemical bonds

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