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...

44. What does
the law of
conservation of
matter
state - - - use
reactants and

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products to
explain answer?

45. Describe the
particles of a
liquid. 46.

Describe
particles of a
gas. 47. What
changes a liquid
to a gas? 48.

How is a mixture
different from a
compound? ...

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MATTER: anything
that has mass
and takes up
space. STATES OF

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MATTER: solid,
liquid or gas.

MASS: measure of
how much matter
is in something.

This is measured
in kilograms

(kg), grams (g)
and milligrams

(mg) and can be
measured using a
balance. **VOLUME:**

the amount of
space something

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takes up.

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Date: _____ Know
the meaning of
the following
terms: atom- the
smallest

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particle of an element that still has the properties of that element.

Chemical

property- a characteristic of matter that can be observed only when matter is changed into a new kind of matter. density-

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how much matter
is in a ...

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help students
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_____ 12. What observations led Czarowski to form his hypothesis? a. how easily penguins propel themselves b. how easily boats propel themselves c. how badly penguins propel themselves d.

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_____ Chapter 11
& 12 Study
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Forces Answer

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Key. Chapter 11:
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(include the
formula. and
circle diagram
for calculating
speed, velocity,
and
acceleration):
Distance: The
length between
two objects or
the length of
the path

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