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The Physical Science Core subject is composed of two major scientific fields, Chemistry and Physics. ... The Writing Team hopes that this Teaching Guide be able to help the teachers in ensuring quality content and strategy in teaching Physical Science. Earth Science is a Core Subject taken in the first semester of Grade 11.

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Physical Science: Senior High School SHS Teaching Guide ...

Physical Science Semester 2

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Define and give Examples of Compounds and Elements

Compounds—Molecules made of more than 1 atom bonded

together. Example: Water (H₂O) Elements—made of only 1

type of atom. Example:

Carbon or Helium Compare and

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*Contrast a Pure Substance
and Mixture (Homogeneous and
Heterogeneous) Pure
Substance: a substance that
contains one element of ...*

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1st Semester Final Exam
Study Guide Physical Science
Student Class Date 1.S8P1.a
The smallest particle of
matter that can be
identified as an element is
called A. an atom. B. a
proton. C. an electron. D. a
molecule. 2. S8P1.a The
chemical formula for methane
is and a model of its
structure is shown. The
model of methane represents
a(n)*

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and Energy Transformations.
Explain transfer of heat
through Conduction,
Convection and Radiation.
Include examples and
diagrams. Explain how*

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potential can change to kinetic energy and vice versa. Give examples of kinetic energy. Give examples of potential energy

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*Physical Science Grade 10
Study Guide Revision When revising, many people find it helpful to write as they work. You are more likely to remember something that you have written than something that you have just looked at in a book. You will also find that you can concentrate better and learn faster if you revise*

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Description.

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Physical Science Basics a.

*What is the formula for
calculating density? _____*

*i. What is the density of a
cube that is 15 g and has a
volume of 3 cm³? _____ ii.*

*Compare the density of solid
H₂O to liquid H₂O, and
explain what observable
effect this has.*

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Study Guide I. Chemistry 1.
What is the difference
between a homogeneous
mixture and a heterogeneous
mixture? 2. List 2 examples
of a homogeneous mixture and
2 examples of heterogenous
mixtures. 3. List 2 examples
of a pure substance and 2
examples of a mixture. 4.*

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Flashcards. Learn. Write.
Spell. Test. PLAY. Match.
Gravity. Created by.
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*class. Terms in this set
(36) Scientific method #1
Ask a Question #2 Do
Background Research #3
Construct a Hypothesis #4
Test Your Hypothesis by
Doing an Experiment*

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