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modified 3 d) Now suppose one
keeps adding 1.0 M HCl to this
buffer until a total of 101 mL of 1.0M

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HCl have been added. At this point, the pH of the solution will be very acidic.

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molecular formula, we must first
find the empirical formula. We start
by imagining a sample of the
compound weighing 100 grams, so
the percentages can be seen as

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grams. 25g of the sample is carbon, 8.3g of the sample is hydrogen, and 66.7g of the sample is oxygen.

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