

# Pogil Calorimetry Answers Heat Energy

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In this POGIL, you will explore how mass, temperature, heat energy, ... 45. Does it take more energy to heat water or ice? Explain your answer, but be careful. The water from the problem underwent a 100°C temperature change, but ice only increased by 50°C. -100-50 0 50 100 1

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to an equality (= sign), we need to introduce a

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reaction in a calorimeter ( $Q$ ) is directly proportional to the mass of the water in the calorime  
(given the symbol  $m$ ) and directly proportional to the change in the temperature of the water

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If 4.168 kJ of heat is added to a calorimeter containing 75.40 g of water, the temperature of water and the calorimeter increases from 24.58 degree C to 35.82 degree C. Calculate the heat capac...

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Calorimetry – Measurement of Heat Energy 7. Determine the mass of a water sample that is heated from an initial temperature of 25 °C to a final temperature of 100. °C following the addition of 1200 J of heat energy. 8. A 100. g sample of pure lead is heated from 10.0°C to

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197.50C by the addition of 3000. J of heat energy.

Calorimetry – Measurement of Heat Energy

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