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Comprehensive in approach, this text explores the scientific principles and methods that underlie the cause

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Fundamentals 440. Another is reduction of an oxidized ion in solution by a so-called redox reaction. 1. and 0 $2 + 4H^+ + 4e^- \rightarrow 2H_2O$. assuming dissociation of water to H^+ and OH^- . Most corrosion reactions do involve water in either the liquid or condensed vapor phases. and is called the electrolyte.

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Electrochemical principles of corrosion. 7. Thermodynamic

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principles of corrosion. 1. Definition of corrosion (1/3) The loss of useful properties of material as a result of chemical or electrochemical reaction with its environment. 1. Definition of corrosion (2/3) ... heat treatment given , e . g . Tempered steel is

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Article: Principles and prevention of crevice corrosion

Principles and Prevention of Corrosion aims at fulfilling these gaps so that the

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reader would know as to how and why the corrosion takes place. It is also useful for practicing engineers as well as design engineers who are concerned about corrosion.

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The mechanism of crevice corrosion. The release of metal ions (M^+) at the anode is balanced by a reaction at a nearby cathodic site, which involves using up oxygen from the surroundings. In a crevice oxygen can

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soon be used up and if the conditions are stagnant enough, replenishment of the oxygen to the crevice from the bulk solution is stifled.

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