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Equation: $\text{MnO}_2 + 4\text{HCl} \rightarrow \text{MnCl}_2 + 2\text{H}_2\text{O} + \text{Cl}_2$ (l) Name the
Substance Oxidised.
(li) Name the
Oxidising Agent. (lii)
Name the Substance
Reduced. (lv) Name
the Reducing Agent.

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