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Chapter 3 ...  
In the Reaction  
Represented by the  
Equation:  $MnO_2 +$***

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**$4\text{HCl} + \text{MnCl}_2 + 2\text{H}_2\text{O}$   
 $+ \text{Cl}_2$  (I) Name the  
Substance Oxidised.  
(II) Name the  
Oxidising Agent. (III)  
Name the Substance  
Reduced. (IV) Name  
the Reducing Agent.**

**Magnesium Chloride  
( $\text{MgCl}_2$ ) - Structure,  
Molecular Mass ...  
NCERT Solutions for  
Class 12 Chemistry**

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**Chapter 3**

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***Magnesium Chloride  
(MgCl<sub>2</sub>) -***

***Magnesium Chloride  
is the chemical  
name of MgCl<sub>2</sub>. Visit  
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