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HOW TO USE The Hazardous Materials Regulations

The atomic number of Al and Cl are 13 and 17, respectively. What will be the number of electrons in Al $3+$ and Cl $-$? Answer: Atomic number of Al = Number of electrons = 13 Number of electrons in Al $3+$ = $13 - 3 = 10$ Atomic number of chlorine = Number of electrons = 17 Number of electrons in Cl $-$ = $17 + 1 = 18$. Question 6.

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