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the conditions of alkene reactions was one of the great puzzles of organic chemistry that was ultimately explained by the theory of aromaticity, which is introduced in Chapter 15. B. Hydration of Alkenes The alkene double bond undergoes reversible addition of water in the presence of moderately concentrated strong acids such as  $\text{H}_2\text{SO}_4$ ,  $\text{HClO}_4$  ...

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STRUCTURE AND ...**

reactions in organic chemistry involve the movement of electron pairs, it follows that these re-actions are but variations on a theme; they are extensions of the simple examples shown in Eqs. 3.2 and 3.5. Thus, mastery of these simple inorganic examples is a necessary first step in understanding the reactions of organic chemistry.

**3.3 REVIEW OF THE CURVED-ARROW NOTATION**

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Write the systematic (IUPAC) names of the following carboxylic acids.

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