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Borates: Handbook of Deposits, Processing, Properties, and ...

Sodium sulfate is a typical electrostatically bonded ionic sulfate. The existence of free sulfate ions in solution is indicated by the easy formation of insoluble sulfates when these solutions are treated with Ba²⁺ or Pb²⁺ salts: Na₂SO₄ + BaCl₂ → 2NaCl + BaSO₄. Sodium sulfate is stable toward most oxidizing or reducing agents.

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