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solvent, substance, ordinarily a liquid, in which other materials dissolve to form a solution. Polar solvents (e.g., water) favour formation of ions; nonpolar ones (e.g., hydrocarbons) do not. Solvents may be predominantly acidic, predominantly basic, amphoteric (both), or aprotic (neither). Organic compounds used as solvents include aromatic compounds and other hydrocarbons, alcohols, esters ...

solvent | Definition, Examples, & Facts | Britannica

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"Sulfate" is the spelling recommended by IUPAC, but "sulphate" was traditionally used in British English.. Structure. The sulfate anion consists of a central sulfur atom surrounded by four equivalent oxygen atoms in a tetrahedral arrangement. The symmetry is the same as that of methane. The sulfur atom is in the +6 oxidation state while the four oxygen atoms are each in the -2 state.

Sulfate - Wikipedia

hydrogen cyanide, also called formonitrile (HCN), a highly volatile, colourless, and extremely poisonous liquid (boiling point 26° C [79° F], freezing point -14° C [7° F]). A solution of hydrogen cyanide in water is called hydrocyanic acid, or prussic acid. It was discovered in 1782 by a Swedish chemist, Carl Wilhelm Scheele, who prepared it from the pigment Prussian blue.

hydrogen cyanide | Description, Uses, Effects, & Facts

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